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which may provide insight into ways to produce that form in the lab. Hydrated iron(III) oxides even when they do not contain iron oxide.

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Iron oxide pigment. Iron oxides are chemical compounds composed of iron and oxygen. All together, there are sixteen known iron oxides and oxyhydroxides. [1]

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Iron oxide has many uses including pigments, ceramics, welding (thermite), along with a wide range of other uses. As with all chemicals, iron oxide should be handled

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The rust you see on metal is an example of natural iron oxide formation. Rust forms when iron is exposed Iron oxides are made in a lab for safety reasons since

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